Evaluation of the efficacy of XP Endo Finisher, Easy Clean, Passive ultrasonic irrigation, and conventional irrigation in the removal of calcium hydroxide in simulated internal root resorption cavities

Avaliação da eficácia do XP-Endo Finisher, Easy Clean, irrigação ultrassônica passiva e irrigação convencional na remoção de hidróxido de cálcio em cavidades de reabsorção internas simuladas

Evaluación de la eficacia de XP-Endo Finisher, Easy Clean, irrigación ultrassónica pasiva e irrigación convencional en la eliminación de hidróxido de calcio en cavidades de reabsorción interna simuladas

Received: 02/16/2021 | Reviewed: 02/22/2021 | Accept: 02/24/2021 | Published: 03/03/2021

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Abstract
This study aimed to evaluate the efficacy of 5 irrigation protocols in removing calcium hydroxide in simulated internal root resorption cavities. Eighteen extracted human teeth, upper anterior and uniradicular, were prepared up to size 50. The teeth were divided longitudinally, and internal root resorption cavities were made by acid demineralization using nitric acid. The same sample was submitted to the five irrigation protocols. The cavities and root canals were filled with Ultracal XS (Ultradent, South Jordan, UT, USA) and stored in an incubator at 37°C for one week. Afterward, the irrigation protocols were carried out: XP-Endo Finisher (XP; FKG, La Chaux-de-Fonds, Switzerland), passive ultrasonic irrigation (PUI), Easy Clean (Bassi/Easy Equipamentos Odontológicos, Belo Horizonte, Brazil) with reciprocating movement (ECR), Easy Clean with continuous rotation (ECC) and conventional irrigation with a syringe (CI). The solutions used, 2.5% NaOCl, and 17% EDTA, were warmed. The total volume of irrigant used for each sample was 20 mL (10 mL of 2.5% NaOCl, 5 mL of 17% EDTA and 5 mL of distilled water). The amount of calcium hydroxide remaining in the internal resorption after irrigation was assessed using scores. The data were analyzed using the Pearson and Kruskal-Wallis (Dunn) tests. Significantly more calcium hydroxide was removed in the XPF group than all other sample groups (P < 0.0001). XPF was superior to the other methods tested to remove calcium hydroxide in teeth with simulated internal resorptions.

Keywords: Dental pulp cavity; Endodontics; Root canal irrigation.
Resumen
Este estudio tuvo como objetivo evaluar la eficacia de 5 protocolos de riego en la eliminación de hidróxido de calcio en cavidades de reabsorción interna simuladas. Se prepararon dieciocho dientes humanos extraídos, anterior superior, y se almacenaron en una incubadora a 37°C durante una semana. Posteriormente, se llevaron a cabo los protocolos de riego: XP-endo Filler (XPF; FKG, La Chaux-De-Fonds, Suiza), riego ultrasonico pasivo (PU), EasyClean (Easy Equipment Odontológicos, Belo Horizonte, Brasil) con movimiento reciprocante (ECR), EasyClean con rotación continua (ECC) e irrigación convencional con jeringa (IC). Las soluciones utilizadas, NaOCl al 2,5% y EDTA al 17%, fueron aquecidas. El volumen total de irrigante utilizado para cada muestra fue de 20 mL (10 mL de NaOCl 2,5%, 5 mL de EDTA 17% y 5 mL de agua destilada). La cantidad de hidróxido de calcio remaneciente en la reabsorción interna después de la irrigación se evaluó por medio de escores. Los datos se analizaron mediante los testes de Pearson y Kruskal-Wallis (Dunn). Significativamente más hidróxido de calcio fue removido en el grupo XPF do que todos los otros grupos (P <0,0001). XPF fue superior a los otros métodos testados para eliminar hidróxido de calcio en dientes con reabsorciones internas simuladas.

Palabras clave: Cavidad pulpar dental; Endodonia; Irrigación del conducto radicular.

1. Introducción

According to its location, root resorption results from the clastic cells activity on mineralized tissues and is classified as internal or external (Patel et al. 2010). Internal root resorption begins within the root canal and requires vital pulp tissue, usually caused by some trauma and tends to be asymptomatic (Majorana et al. 2003). When detected early, treatment is often successful, and the prognosis is favorable (Gabor et al. 2012). Endodontic treatment is the treatment of choice for this pathological condition (Haapasalo 2006), but the resorption cavity's irregular borders pose technical difficulties in preparing the canal and its obturation (Haapasalo 2006).

Calcium hydroxide (Ca(OH)$_2$) is used as an intracanal medication to treat internal root resorption when it is impossible to treat it in only one session (Mohammadi & Dummer, 2011). The removal of Ca(OH)$_2$ before the obturation is recommended because remnants retained in these extensions may interfere with the penetration and adhesion of sealers in dentinal tubules (Calt & Serper, 1999) and also with their physical properties (Margelos et al. 1997). A frequently described method for removing Ca(OH)$_2$ from the root canal is root canal instrumentation in combination with irrigation with NaOCl and EDTA (Salgado et al. 2009, Keskin et al. 2017, Dioguardi et al. 2018). However, previous studies have demonstrated the ineffectiveness of conventional irrigation techniques in removing Ca(OH)$_2$ from root canal irregularities (Keskin et al. 2017, Topcuoglu et al. 2015, Wigler et al. 2017).

Alternative techniques of irrigating agitation are necessary for effective root canal cleaning (Fernandes et al. 2020).
Among the available techniques we chose three for this study: 1- Passive ultrasonic irrigation (PUI), a technique that consists of ultrasonic tips activation inside the canal, promotes energy release by cavitation and acoustic microstreaming of the liquid (van der Sluijs et al. 2007, Jiang et al. 2011, Macedo et al. 2014). 2- An instrument made of a plastic called EasyClean (25.04) (Bassi/Easy Equipamentos Odontológicos, Brazil) was developed for promoting the cleaning of the apical third by rotating or reciprocating agitation (Kato, et al 2016, Duque et al. 2017). 3- A nickel-titanium file (XP-endo Finisher; FKG, La Chaux-de-Fonds, Switzerland) with a size #25, non-tapered instrument, designed to reach difficult-to-access areas (alternating martensite and austenite phases according to temperature) adapts to the root canal in a three-dimensional way (Keskin C et al. 2017).

Based on evidence that better results can be obtained in removing intracanal Ca(OH)\textsubscript{2} using agitation protocols for irrigants (Keskin et al. 2017, Topcuoglu et al. 2015, Wigler et al. 2017) it is crucial to evaluate these protocols in teeth with simulated internal resorption. We considered the null hypothesis that different irrigation methods are equally effective in removing Ca(OH)\textsubscript{2}.

2. Methodology

The local ethics committee approved this study (No. 2.485.437). Eighteen extracted human teeth, upper anterior with single straight root, showing complete root formation, were selected. Teeth with internal or external resorption, cracks or fractures, previous endodontic treatment, curvatures more outstanding than 15°, and calcified canals were excluded. A radiographic examination confirmed the presence of a single root canal. The specimens were stored in 0.1% thymol (Lenzafarm, Belo Horizonte, Brazil). The crowns were removed to standardize the root length to 20 mm. After preparing the access cavities using spherical diamond tips, the working lengths (WL) were defined by introducing a #15 K-file (Maillefer-Dentsply, Ballaigues, Switzerland) until its tip was visible in the foramen and by subtracting 1 mm from that measurement. The canals were instrumented by using Reciproc system (VDW, Munich, Germany) up to size R50 (50.05) under irrigation with 5 mL of 2.5% NaOCl (Lenzafarm, Rio De Janeiro, Brazil) delivered with a 30-gauge needle (NaviTip; Ultradent, South Jordan, UT, USA) positioned 1 mm short of WL. After completing mechanical preparation, the canals were irrigated with 5 mL of distilled water and dried with paper points.

The specimens were embedded in silicone impression material (Zetaplus; Zhermack, Rovigo, Italy) placed in 1.5 mL Eppendorf tubes. After the silicone set, the specimens were removed, and longitudinal grooves were prepared along with the roots on buccal and lingual surfaces. The teeth were then cleaved along the long axis.

Preparation of internal root resorption cavities

A cylindrical chamber (2 mm in diameter and 2 mm in height) was made in the root middle third with compound resin to limit the substances’ application, and the dentin was covered by the application of an adhesive system and polymerization. A demineralization protocol induced internal resorption with 5% nitric acid and 8% sodium hypochlorite. Samples were kept at -1°C (±3°C) during the internal resorption induction period. At the end of 3 cycles, the chamber was detached, and residual substances were removed with a 24h wash in running water (da Silveira et al. 2014).

Before Ca(OH)\textsubscript{2} (Ultracal XS, Ultradent) insertion into the teeth, the roots were subjected to ultrasonic baths with 17% EDTA (Lenzafarm), running water, 2.5% NaOCl and running water again, all for one minute each, and dried with paper points. For insertion the Ca(OH)\textsubscript{2}, we used a 29-gauge NaviTip needle (Ultradent). The root halves were reassembled in their molds, and the access cavities were sealed with temporary fillings (Villevie, Joinville, Brazil). The specimens were then stored at 37°C and 100% humidity for one week in an incubator. The temporary fillings were then removed. A #50 K-file (Maillefer-Dentsply, Ballaigues, Switzerland) was introduced into the WL canal to loosen the paste and create a space for irrigation.
needle. Five teeth were used as a negative control and were not filled with Ca(OH)₂. Five teeth were used as a positive control to ensure that there was no medication loss during transport and handling, and they were filled with Ca(OH)₂, but no removal procedure was performed. The same sample was subjected to the five irrigation protocols, and before being reused, the teeth were subjected to the abovementioned ultrasonic baths.

**EasyClean reciprocating (ECR)**

The sample was irrigated with 5 mL of 2.5% NaOCl, 5 mL of 17% EDTA and again 5 mL of 2.5% NaOCl using a 30-gauge needle and syringe (Ultradent) and agitated with EasyClean using the VDW Silver motor (VDW, Munich, Germany) in the Reciproc All mode. The instrument was operated at 1 mm short of WL for 1 min (3 cycles of 20 s) with each solution.

**EasyClean continuous rotation (ECC)**

The same technique described in the previous group was used, but EasyClean was operated at low speed (Dabi Atlante, Ribeirão Preto, Brazil), at about 20,000 rpm at 1 mm short of WL.

**Passive ultrasonic irrigation (PUI)**

Irrigation was performed similarly as in the previous groups but was passively agitated using a piezoelectric ultrasonic unit (Gnatus, Ribeirão Preto, Brazil) with an Irrisonic tip (Helse, Santa Rosa do Viterbo, Brazil) positioned 1 mm short of WL without touching the walls. The apparatus was used at 20% power by performing three cycles of 20 s (1.7 mL each cycle) with 5 mL of 2.5% NaOCl, followed by another three cycles of 20 s with 5 mL of 17% EDTA, ending with three cycles of 20 s with 5 mL of 2.5% NaOCl.

**XP-endo Finisher (XPF)**

XP-endo Finisher (FKG, La Chaux de Fonds, Switzerland) was used in an electric endodontic motor (VDW Silver) and cooled down to be removed from the plastic tube. Root canals were irrigated with 5 mL of 2.5% NaOCl using a 30-gauge needle and syringe. The XPF file was inserted into the root canal and operated at a speed of 800 rpm and torque of 1N/cm for 3 cycles of 20s with vertical movements 1 mm short of WL. Next, the same procedure was then performed using 5 mL of 17% EDTA, and another irrigation with 5 mL of 2.5% NaOCl with agitation was performed for three cycles of 20s. The solutions were renewed every 20s cycle and warmed to 38°C to allow the XPF to assume a hook shape in its last 10 mm.

**Conventional irrigation (CI)**

Irrigation was performed similarly as in the previous groups using a 30-gauge needle (Ultradent) positioned 1 mm short of WL. No irrigant agitation was performed.

At the end of the protocols, each sample was irrigated with 5 mL of distilled water. The total irrigant volume for each sample was 20 mL (5 mL of 2.5% NaOCl + 5 mL of 17% EDTA + 5 mL of 2.5% NaOCl + 5 mL of distilled water). The solutions were warmed to 38°C in all groups. An endodontic specialist performed all the procedures.

The root halves were separated, and digital images of each half's internal root resorptions were obtained using a stereomicroscope at 20x magnification. The sequence of all images was randomized, and two examiners blinded to group assignment evaluated each image. The amount of Ca(OH)₂ remaining in the internal resorption cavity after irrigation was scored using the following classification described by Van der Sluis et al. 2007: 0 - cavity is empty, 1 - less than half of cavity is filled with Ca(OH)₂, 2 - more than half of cavity is filled with Ca(OH)₂, 3 -cavity is completely filled with Ca(OH)₂. (Figure 1)
**Figure 1.** Representative images of scores. A) Score 0: cavity is empty. B) Score 1: less than half of cavity is filled with Ca(OH)$_2$. C) Score 2: more than half of cavity is filled with Ca(OH)$_2$. D) Score 3: cavity is completely filled with Ca(OH)$_2$.

Source: Authors.

The Shapiro-Wilk revealed a non-normal distribution of data, so the Kruskal-Wallis (Dunn) test at a 5% level of significance was used to compare the groups tested. Statistical analyses were done using the Biostat 4.0 program.

**3. Results**

The inter-examiner agreement was 0.8177 (P < 0.0001), as determined by the Pearson correlation test. Tables 1 and 2 detail the distribution of scores for all groups. Positive control specimens confirmed that Ultracal XS was not lost from the wells during disassembly and transport. The Kruskal-Wallis (Dunn) test showed that there was a statistically significant difference in Ca (OH)$_2$ removal scores between the groups tested (P < 0.0001). XPF removed significantly more Ca (OH)$_2$ than did all other groups tested (P < 0.0001). There was no statistically significant difference between CI and ECR (P > 0.0001).
Table 1. Distribution and percentage of scores of groups tested.

<table>
<thead>
<tr>
<th>Groups</th>
<th>teeth/halves</th>
<th>0 (%)</th>
<th>1 (%)</th>
<th>2 (%)</th>
<th>3 (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control -</td>
<td>5/10</td>
<td>10 (100)</td>
<td>0 (0)</td>
<td>0 (0)</td>
<td>0 (0)</td>
</tr>
<tr>
<td>Control +</td>
<td>5/10</td>
<td>0 (0)</td>
<td>0 (0)</td>
<td>0 (0)</td>
<td>10 (100)</td>
</tr>
<tr>
<td>ECC</td>
<td>18/36</td>
<td>0 (0)</td>
<td>3 (8.3)</td>
<td>19 (52.7)</td>
<td>14 (38.8)</td>
</tr>
<tr>
<td>ECR</td>
<td>18/36</td>
<td>0 (0)</td>
<td>0 (0)</td>
<td>8 (22.2)</td>
<td>28 (77.7)</td>
</tr>
<tr>
<td>CI</td>
<td>18/36</td>
<td>0 (0)</td>
<td>1 (2.7)</td>
<td>2 (5.5)</td>
<td>33 (91.6)</td>
</tr>
<tr>
<td>PUI</td>
<td>18/36</td>
<td>4 (11.1)</td>
<td>16 (44.4)</td>
<td>10 (27.7)</td>
<td>6 (16.6)</td>
</tr>
<tr>
<td>XPF</td>
<td>18/36</td>
<td>27 (75)</td>
<td>9 (25)</td>
<td>0 (0)</td>
<td>0 (0)</td>
</tr>
</tbody>
</table>

Source: Authors.

Table 2. Minimum values (MN), maximum values (MX), medians (MD), interquartile deviations (ID) and statistical analysis using the Kruskal-Wallis (Dunn) test for amount of remaining Ca(OH)₂ of the experimental groups, according to the classification of Van der Sluis et al. (2007).

<table>
<thead>
<tr>
<th></th>
<th>ECR</th>
<th>ECC</th>
<th>PUI</th>
<th>XP</th>
<th>IC</th>
<th>P</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>N</strong></td>
<td>36</td>
<td>36</td>
<td>36</td>
<td>36</td>
<td>36</td>
<td></td>
</tr>
<tr>
<td><strong>MN</strong></td>
<td>2</td>
<td>1</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>1</td>
</tr>
<tr>
<td><strong>MX</strong></td>
<td>3</td>
<td>3</td>
<td>3</td>
<td>1</td>
<td>3</td>
<td>P &lt;0.0001</td>
</tr>
<tr>
<td><strong>MD</strong></td>
<td>3.00</td>
<td>2.00</td>
<td>1.00</td>
<td>0.00</td>
<td>3.00</td>
<td></td>
</tr>
<tr>
<td><strong>(ID)</strong></td>
<td>(0.00)AA</td>
<td>(1.00)A</td>
<td>(1.00)AB</td>
<td>(0.25)B</td>
<td>(0.00)AB</td>
<td></td>
</tr>
</tbody>
</table>

Source: Authors.

4. Discussion

Ca(OH)₂ can be used as an intracanal medication to treat internal root resorption because of its ability to increase pH, control bleeding and promote residual pulp necrosis (Haapasalo et al. 2006). Previous studies have used burs to simulate internal root resorption (Keskin et al. 2017, Topcuoglu et al. 2015). This technique advantage is the standardization of the cavities, but these cavities show regular borders to compare natural resorptions, which are irregular lesions. In the present study, the root resorptions were induced by acid demineralization, and this method produces irregular cavities, very close to the clinical situation in vivo (da Silveira PF et al. 2014). The same sample was submitted to the five irrigation protocols to work
around the fact that with cavities induced by acid demineralization, it would not be possible to control these cavities' depth. Therefore, the same cavities were analyzed in all protocols (Calt Set al. 1999, Margelos et al. 1997, Barbizam et al 2008, Ghabraei et al. 2017).

Besides the instrument's physical action, the irrigant's chemical and mechanical properties also play an essential role in the removal of Ca(OH)$_2$ (Salgado et al. 2009). A previous study showed that the use of 15% EDTA or NaOCl alone could not remove Ca(OH)$_2$ from the root canal; however, the combination of the two irrigants improved removal efficacy (Margelos et al. 1997). In the present study, similarly to previous studies, NaOCl and EDTA's combined use was tested (Keskin et al. 2017).

Previous studies have demonstrated the ineffectiveness of conventional irrigation techniques in removing Ca(OH)$_2$ from root canal irregularities (Keskin et al. 2017, Topcuoglu et al 2015, Wigler et al. 217, Ghabraei et al.2017). Thus, in the present study, additional steps using XPF, ECC, ECR, PUI, and CI were tested to remove Ca(OH)$_2$ from the resorption cavities. The null hypothesis that there is no significant difference between the groups tested was rejected.

This study's main finding was that the XPF instrument gave the best results for Ca(OH)$_2$ removal in the simulated resorption cavities. ECR and CI had the worst performance without significant statistical differences between them.

The XPF instrument was developed to access irregular areas of the root canal system and is suggested by the manufacturer to remove intracanal medication. Wigler et al. 2017 suggested that irrigation longer than one minute with XPF should be tested. In the present study, we used the following in all protocols: 2.5% NaOCl, 17% EDTA, and distilled water totaling 20 mL of irrigating solution, and three min of agitation. All solutions were warmed to 38°C. Above 35°C, the XPF instrument converts to the austenitic form and expands, and a solution at a lower temperature could adversely affect the results of that protocol in “in vitro” studies. Keskin et al. 2017 stirred the solutions for two min, without warming them with a 15 mL total volume of irrigant. As a result, only 25% of the cavities were utterly cleaned. In our study, Ca(OH)$_2$ was removed entirely in 75% of the cavities. Thus, XPF combined warmed NaOCl and EDTA solutions and agitation for three min increased the Ca(OH)$_2$ removal from simulated internal root resorption cavities. Warming solutions seems to be of great value in obtaining better results in in vitro studies. The fact that XPF showed better results could also be explained by its expansion capacity, allowing this instrument to access the irregular resorption cavities spaces’, favoring the displacement and removal of Ca(OH)$_2$.

These results agree with a recent study that also showed the XPF protocol’s superiority in the removal of organic tissues from the interior of simulated internal resorption cavities (Ulusoy et al. 2018).

In the present study, PUI showed only 11.11% of cavities free of Ca (OH)$_2$. These results contrast with previous studies that reported more than 40% Ca(OH)$_2$ free cavities (Keskin et al. 2017, Dioguardi M et al. 2018). This can be explained by the difference in power used in the ultrasonic unit, which was 20% in the current work and 60% in the previous one. Higher power leads to a more significant motion’s range of the ultrasonic tip and greater cavitation and acoustic microstreaming (Ahmad et al. 1987), contributing to more effective removal of Ca(OH)$_2$. In the PUI group, the insert's tip was kept static in the apical third of the canal. It is known that the phenomena of cavitation and acoustic microstreaming are more intense at the insert tip; if it is used with vertical movements along with the canal, allowing the tip to vibrate at the same level as resorption, the results could be better (Vivan et al. 2016).

The EasyClean instrument had not yet been tested for its ability to remove Ca(OH)$_2$ in teeth with simulated internal root resorption cavities. In the present study, the ECC group gave better results when compared to the ECR and CI groups. This can be explained by the higher speed used with this instrument, resulting in a greater agitation of the liquids inside the root canal, which favors the displacement of Ca(OH)$_2$. These results corroborate an earlier study that also tested EasyClean with reciprocating and continuous rotation movements to remove debris from canals and isthmuses (Duque JÁ et al. 2018).
The current study showed that the worst results were obtained in the ECR and CI groups, agreeing with previous studies that reported CI's ineffectiveness in removing Ca(OH)$_2$ (van der Sluis LW et al. 2007, Tasdemir T et al. 2011, Yucel AC et al. 2013). The poor performance of ECR can be attributed to its inability to create a cavitation effect.

5. Conclusion

The methods were tested in simulated resorption cavities in straight root canals. Further studies are required to evaluate these protocols' efficacy in simulated resorption cavities made in curved root canals, using the acid demineralization and with possible diffusion of Ca(OH)$_2$ in the dentinal tubules. In conclusion, the XP-endo Finisher file method was superior to the other methods used to remove Ca(OH)$_2$ in teeth with simulated internal resorption cavities.

References


